

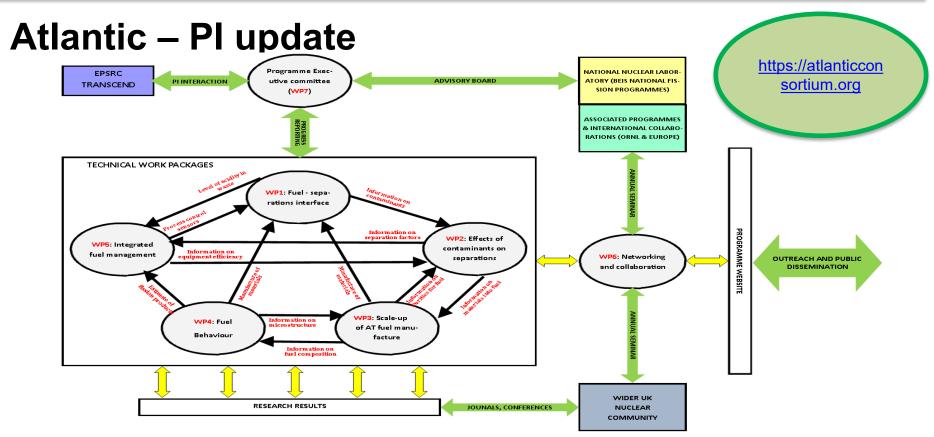


# ATLANTIC: Accident ToLerANT fuels In reCycling

PI – Prof. Bruce Hanson Nuclear Academics Discussion Meeting 7<sup>th</sup> to 8<sup>th</sup> September 2021 Cambridge Advanced Technology

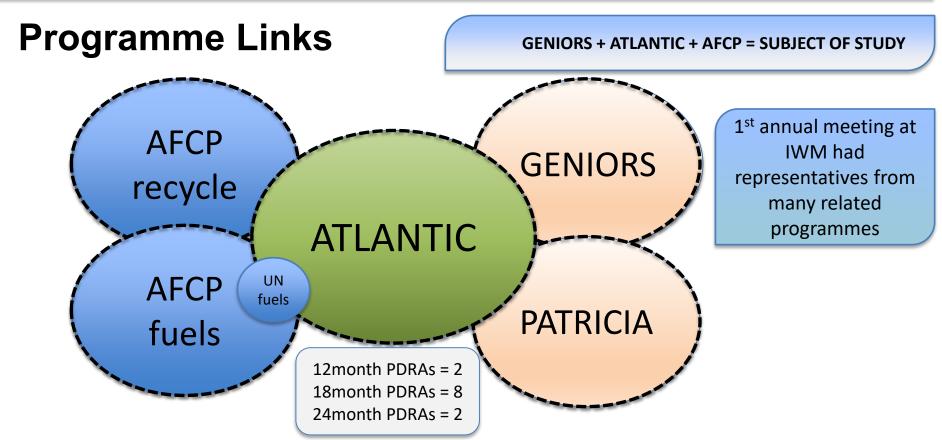
















# Atlantic – technical highlights

•WP1 – Julio Vasquez-Chavez (UoL) is investigating voloxidation of fuels as a pretreatment for reprocessing; using air and steam on Zr. At >900°C the cladding forms a brittle oxide.

•WP2 - **Steve Faulkner (Oxford)** is measuring speciation by deconvolution. Developing methods to separate signals using time-, wavelength-, and temperature dependent luminescence spectroscopy.

•WP3 - **Rob Harrison (UoM)** working on Ce<sub>3</sub>Si<sub>2</sub> oxidation as U<sub>3</sub>Si<sub>2</sub> surrogate. TGA, XRD, HRTEM, STEM-EDS and EFTEM have confirmed the formation of CeO<sub>2</sub>, SiO<sub>2</sub> and Si up to 750°C in air.

•WP4 - **Eleanor Lawrence Bright (UoB)** is characterising corrosion and oxidation of UN surfaces using TEM, XRR, XPS. UN surface passivates at room temperature ( $U_2N_3$  interlayer forms).

•WP5 - **Ilka Schmueser (UoE)** is developing electrochemical sensors for process control. Use electrochemistry to generate a signal that tells you something about a target chemical.



WP1.3 UN Fuel Inventory at Higher Burnups **Using FISPIN** 

## Lancaster 🍱 University

#### Work carried out in AFCP Fuels Programme

Attempting to create the first UN SIMFuel • that replicates the chemical properties of thi material after removal from a reactor.

#### Model inventories used for the thermodynamic calculations

In spent fuel with the highest BU (60 MW d kg<sup>-1</sup> the solid solution comprises:

- 87 mol% UN, ٠
- Transuranics nitrides (AnN) 1.4 mol% (with 1.2 % ٠ PuN),
- Lanthanide nitrides (LnN) 2.5 mol% (with 1 mol% ٠ NdN)
- Transition metals (TrN) 1.7 mol%. ٠

nis	Busine	ment for NAT ss, Energy strial Strategy	Advanced Fuel Cycle Programme	
	#	BURNUPS (GWd/tU)	IRRADIATION TIMES (YEARS)	Nitride and silicide SIMFUELS
	1	5	1⁄2	(AFCP)
-1)	2	10	1	
-)	3	15	11/2	
	4	20	2	
	5	25	21/2	
	6	30	3	
	7	35	31⁄2	Dissolution
	8	40	4	Trials (ATLANTIC)
	9	45	41⁄2	
	10	50	5	
	11	55	51/2	



### **WP2.2** Practical Statistical Modelling of Spent

### **Fuel Compositions**

MANCHESTER 1824

The University of Manchester

Create a dataset for Gen III(+) reactor systems (e.g. EPR) for ATLANTIC SF targets

- To support our separations work, various key parameters required for unknown SNF compositions:
  - Elemental (g/tHM)
  - Decay Heat (W/tHM)
- Both can be derived from isotopic concentrations
- Calculate from basic input parameters:
  - Initial Enrichment (%235U or %Pu)
  - Burnup (GWd/tHM)
  - Post- Reactor Cooling Time (y)

Isotopes	modelled
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Light Fission Products	Heavy Fission Products	Actinides		
<sup>4</sup> He	<sup>117</sup> Sn, <sup>118</sup> Sn, <sup>119</sup> Sn, <sup>120</sup> Sn, <sup>122</sup> Sn, <sup>124</sup> Sn, <sup>126</sup> Sn	<sup>234</sup> U, <sup>235</sup> U, <sup>236</sup> U, <sup>238</sup> U		
<sup>77</sup> Se, <sup>78</sup> Se, <sup>79</sup> Se, <sup>80</sup> Se, <sup>82</sup> Se	<sup>121</sup> Sb, <sup>123</sup> Sb, <sup>125</sup> Sb	<sup>237</sup> Np		
<sup>81</sup> Br	<sup>125</sup> Te, <sup>126</sup> Te, <sup>128</sup> Te, <sup>130</sup> Te	<sup>238</sup> Pu, <sup>239</sup> Pu, <sup>240</sup> Pu, <sup>241</sup> Pu, <sup>242</sup> Pu		
<sup>82</sup> Kr, <sup>83</sup> Kr, <sup>84</sup> Kr, <sup>85</sup> Kr, <sup>86</sup> Kr	<sup>127</sup>  , <sup>129</sup>	<sup>241</sup> Am, <sup>242m</sup> Am, <sup>243</sup> Am		
<sup>85</sup> Rb, <sup>87</sup> Rb	<sup>128</sup> Xe, <sup>130</sup> Xe, <sup>131</sup> Xe, <sup>132</sup> Xe, <sup>134</sup> Xe, <sup>136</sup> Xe	<sup>242</sup> Cm, <sup>243</sup> Cm, <sup>244</sup> Cm, <sup>245</sup> Cm, <sup>246</sup> Cm		
<sup>86</sup> Sr, <sup>88</sup> Sr, <sup>89</sup> Sr, <sup>90</sup> Sr	<sup>133</sup> Cs, <sup>134</sup> Cs, <sup>135</sup> Cs, <sup>137</sup> Cs			
<sup>89</sup> Y	<sup>134</sup> Ba, <sup>135</sup> Ba, <sup>136</sup> Ba, <sup>137</sup> Ba, <sup>138</sup> Ba			
90Zr, 91Zr, 92Zr, 93Zr, 94Zr, 95Zr, 96Zr	<sup>139</sup> La			
<sup>95</sup> Nb	<sup>140</sup> Ce, <sup>141</sup> Ce, <sup>142</sup> Ce, <sup>144</sup> Ce			
<sup>95</sup> Mo, <sup>96</sup> Mo, <sup>97</sup> Mo, <sup>98</sup> Mo, <sup>100</sup> Mo	<sup>141</sup> Pr			
<sup>99</sup> Tc	<sup>142</sup> Nd, <sup>143</sup> Nd, <sup>144</sup> Nd, <sup>145</sup> Nd, <sup>146</sup> Nd, <sup>148</sup> Nd,			
	<sup>150</sup> Nd			
<sup>100</sup> Ru, <sup>101</sup> Ru, <sup>102</sup> Ru, <sup>103</sup> Ru, <sup>104</sup> Ru, <sup>106</sup> Ru	<sup>147</sup> Pm, <sup>148m</sup> Pm			
<sup>103</sup> Rh	<sup>147</sup> Sm, <sup>148</sup> Sm, <sup>149</sup> Sm, <sup>150</sup> Sm, <sup>151</sup> Sm, <sup>152</sup> Sm,			
	<sup>154</sup> Sm			
<sup>104</sup> Pd, <sup>105</sup> Pd, <sup>106</sup> Pd, <sup>107</sup> Pd, <sup>108</sup> Pd, <sup>110</sup> Pd	<sup>153</sup> Eu, <sup>154</sup> Eu, <sup>155</sup> Eu			
<sup>109</sup> Ag, <sup>110m</sup> Ag	<sup>154</sup> Gd, <sup>156</sup> Gd, <sup>158</sup> Gd, <sup>160</sup> Gd			
<sup>110</sup> Cd, <sup>111</sup> Cd, <sup>114</sup> Cd	<sup>159</sup> Tb			
<sup>115</sup> In				



### WP3.2 Synthesis of uranium nitride

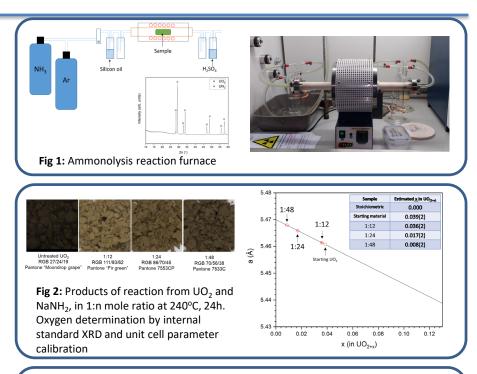


#### Synthesis by reaction with ammonia gas

- An ammonolysis reaction furnace has been designed, tested and operated successfully – Fig 1.
- Ammonolysis of UF<sub>4</sub> proved partially successful, yielding phase assemblage of UN<sub>2</sub> and UO<sub>2</sub>,
- Exploring alternate route starting from NH<sub>4</sub>UF<sub>8</sub> precursor

#### Synthesis by low temperature with NaNH<sub>2</sub> molten salt

- Method used to synthesise transition metal nitrides from oxides by reaction in NaNH<sub>2</sub> molten salt at ca. 240°C for 24 h.
- Reaction between uranium oxides and NaNH<sub>2</sub> investigated; failed to form nitride phase, but reaction of UO<sub>2+x</sub> with NaNH<sub>2</sub> effects a low temperature reduction to stoichiometric UO<sub>2</sub>



#### Recent publications from ATLANTIC and PACIFIC:

- A. Mason *et al.*, molten salt synthesis of Ce doped zirconolite for the immobilisation of pyroprocessing wastes and separated plutonium, Ceramics International, *in press*.
- S. Sun *et al.*, On the existence of the compound "Ce<sub>3</sub>NbO<sub>7+6</sub>" prepared under air atmosphere, Journal of Rare Earths, *in press*.



WP4.2 Fabrication and testing of uranium nitride thin films

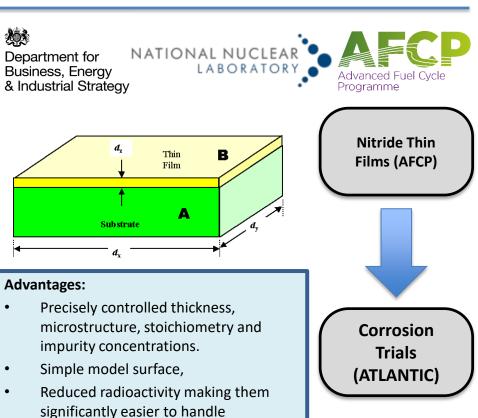


#### Work carried out in AFCP Fuels Programme

 Fabrication of poly epitaxial and single crystal thin films of UN and U<sub>2</sub>N<sub>3</sub>

#### Next stage

- Steam rig experiments Novoclave at 500 °C, 500 bar
- Dopants co-deposition within the films, e.g. Cr to investigate the effect on corrosion resistance
- Fuel-cladding interaction Deposition directly onto cladding materials
- Comparison between nitrides, oxides and silicides





WP5.1 Sensor placement and quantification of uncertainty of readings



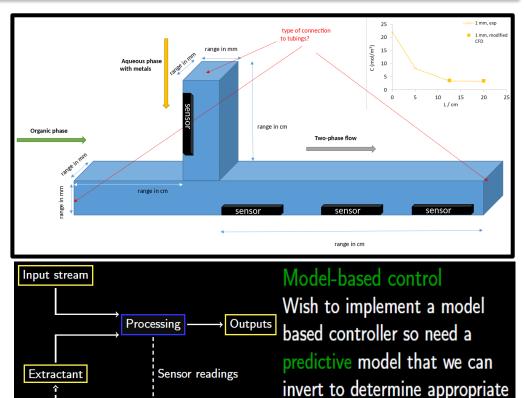
changes in operation.

Determine number and placement of sensors in intensified operating units to enable

- Monitoring of performance for quality assurance and safety and
- Real time control of the processing.

Wish to design and operate an intensified processing unit:

- **Decision variables:** size, configuration, extractant, sensors, flow regime, . . .
- Measured variables: concentration in aqueous phase, flow regime ?
- Criteria for optimization: economic, number of sensors, performance, safety



Controller





## Atlantic – project status

	University		In Post	Started
WP1 Fuel-Separations interface				
WP 1.1 Voloxidation as a pre treatment for accident tolerant fuels	Leeds	PhD		
WP 1.2 The effect of scale up of dissolution kinetics	Leeds	PDRA		
WP 1.3 Corrosion and Dissolution of Accident Tolerant Fuels under Conditions Relevant to Head End	Lancaster	PDRA		
WP 1.4 Molecular Simulation of the Corrosion of Accident Tolerant Fuels: A Modelling Study	Lancaster	PhD		
WP2 Effects of Contaminants on Separations				
WP 2.1 Manufacture of functionalised BTPhen ligands	Reading	PDRA		
WP 2.2 Radiation Stability Testing of Ligands	Manchester	PDRA		
WP 2.3 Development of new tools to analyse speciation	Oxford	PDRA		
WP3 Investigation and Optimisation of Accident Tolerant Fuel Materials				
WP 3.1 Manufacture, Characterisation and Testing of Uranium Nitride Fuels	Sheffield	PDRA		
WP 3.2 Manufacture, Characterisation and Testing of Uranium Silicide Fuels	Manchester	PDRA		
WP 3.3 Radiation Effects in Novel Accident Tolerant Fuels	Liverpool	PDRA		
WP 3.3 Accident Tolerant Nuclear Fuels – Options and Designs (tbc)	Liverpool	PhD		
WP4 Fuel Behaviour: non-stocihiometry and the fuel-water interface				
WP 4.1 High Resolution NMR analysis of Uranium Silicides and Nitrides	Cambridge	PDRA		
WP 4.2 Corrosion Tests of Thin Film Uranium Silicides and Nitrides	Bristol	PDRA		
WP 4.3 Characterisation and Oxidation of Uranium Silicide Phases	Bristol	PhD	•	
WP 4.4 Atomic scale modelling of UN fuel	Imperial	PDRA		
WP 4.5 Ab initio random structure searching to improve fabrication routes for U3Si2 and UN fuels	Cambridge	PhD		
WP5 Integrated Management of Accident Tolerant Fuels				
WP 5.1 Development of High Efficient Separation Technologies	UCL	PDRA		
WP 5.2 Sensor Development and Optimal Placement	Edinburgh	PDRA		

### Mid Term Review 25<sup>th</sup> March



# Conclusions

How has the research landscape, and therefore the role of ATLANTIC within it, changed since funding was awarded?

The proposal was submitted on the 28<sup>th</sup> March 2018 = AFCP + COVID + Brexit Deal on EU Research

Aim to provide a clear view on the Accident tolerant fuels and claddings (ATFC) technology of choice

- From a <u>fundamental science perspective</u> = proof of concept on manufacture of uranium nitride and silicide fuels;
- From a <u>technological perspective</u> = work so far (fuels) has raised the TRL to 2-3 and AFCP is taking over;
- From a <u>strategic perspective</u> = too early to call, but through AFCP Quarterly Meetings, BEIS are aware of ATLANTIC and its aims





# Acknowledgements



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